Volume 14 • 2015

ISSN 1876-6196



Procedia Chemistry

2nd Humboldt Kolleg in Conjunction with International Conference on Natural Sciences 2014, HK-ICONS 2014

Editors:

Roy Hendroko Setyobudi, Hugo Scheer, Leenawaty Limantara, Yuzo Shioi, Leszek Fiedor, Tatas H.P. Brotosudarmo and Monika N.U. Prihastyanti

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Procedia Chemistry 14 (2015) i

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Nomenclature

$\Delta t_{\rm R}$	retention ti
R	retention ti
R Ratio	ratio betwe
v/v	solvents vo

1. Introduction

Chromatography m separation¹. Since th chromatography^{1,2}, col liquid chromatography conducting researches in the development of four main parameters, the best conditions for Pigment analyses of lea

In main separation separation occurs dur HPLC/UFLC columns separating pigments ha of small-sized skeletor with particulate packin performance in separat good separation and st separation. Two most silica. There are number photosynthetic pigment

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2nd Humboldt Kolleg in conjunction with International Conference on Natural Sciences, HK-ICONS 2014

Separation of Photosynthetic Pigments by High-Performance Liquid Chromatography: Comparison of Colum Performance, Mobile Phase, and Temperature

Indriatmoko^a, Yuzo Shioi^a, Tatas Hardo Panintingjati Brotosudarmo^a, Leenawaty Limantara^a*

^aMa Chung Research Center for Photosynthetic Pigments, Universitas Ma Chung, Malang 65151, East Java, Indonesia

Abstract

High-performance liquid chromatography (HPLC) has been commonly used as method of separating and identify photosynthetic pigments such as chlorophylls and carotenoids because of such advantages as speed, high resolution at sensitivity. In this technique, high separation relies largely on the type of column material. This study compared the efficiency five reverse-phase columns, C8, C18, C18 monolithic, π -NAP, and cholester, for separation of photosynthetic pigments at seven fixed conditions of mobile phase and temperature. This investigation also analysed the parameters of Δt_R and t_R ratio for select pigments and resolution for structural isomers, such as α - and β -carotene. Among above columns tested, cholester column suitable for separation of pigments not only for a broad range of polarity, but also for hydrophobic pigments in a simple motif phase. This finding can help in the selection of column and HPLC parameters in separating photosynthetic pigments.

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Keyword: Cholesteryl bonded; HPLC column; monolithic packing; particulate packing; photosynthetic pigments; reverse phase.

*Corresponding author. Tel: +62 813 2636 0303, Fax+62 341 550 175 *E-mail address:* leenawaty.limantara@machung.ac.id

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nents; reverse phase.

Indriatmoko et al. / Procedia Chemistry 14 (2015) 202 - 210

Nomenclature

tR

MR retention time difference retention time t_R Ratio ratio between retention time of two pigment peaks v/v solvents volume ratio

1. Introduction

Chromatography method has been introduced since 1905 as specialized technique for photosynthetic pigments sparation¹. Since then, several methods have been developed and commonly used, e.g., thin-layer chromatography^{1,2}, column chromatography^{3,4}, and high-performance liquid chromatography (HPLC)⁵⁻ . Ultra-fast laud chromatography (UFLC) was one of the newest generations of HPLC which provide special advantages in anducting researches with low time consuming and high resolution data^{8,9}. These advantages gave an opportunity in the development of a low cost and rapid analysis method. Generally, quality of UFLC separation is affected by four main parameters, i.e., mobile phase, flow rate, column temperature, and column type. Thus, an exploration of the best conditions for pigment separation had become a challenge for chromatography researchers in the world. Pignent analyses of leaves of higher plants were reported using different UFLC/HPLC analytical methods¹⁰⁻¹²

In main separation parameters, column material has been understood as an important part where pigment separation occurs during analysis. Other parameters are usually set depend on column type. Generally, HPLC/UFLC columns are distinguished as monolithic and particulate packing types¹³. Ability of these columns in sparating pigments had reported for various samples and their improvements^{5,14–16}. Monolithic column is consisted of small-sized skeletons and wide through-pores which can be achieved higher separation efficiency than the case with particulate packing columns at a similar pressure drop¹⁷. There are several reports on the monolithic column reformance in separating photosynthetic pigments^{16,18,19}. This column type is known for its advantages in providing good separation and short time analysis²⁰. Particulate packing columns have also been widely used for pigment eparation. Two most well-used particulate packing column are octyl (C8) and octadecyl (C18) types based on slica. There are numbers of reports on these C8 and C18 which used to develop optimized method for analysis of photosynthetic pigments^{5,6,14,15}

Recently, new types of column based on napthylethtyl bonded silica packing and cholesteryl bonded silica racking were invented. Cholester column is basically similar with conventional ODS column as their equivalent hydrophobicity. Nevertheless, cholester column has high sensitivity for hydrophobic compound due to their strong stree-selectivity. Like cholester column, π -NAP column has unique specific selectivity in separation. This mphylethyl bonded silica packing column was built for π - π interactions for hydrophobic compound. These advantages, in case of photosynthetic pigments separation, provide better chance to provide good separation of carotenes group.

In previous investigation, two silica particulate packing columns (C18 and C8) were analysed as the standard in addition to C18 monolithic type column to understand the effect of carbon chain length and the difference between particulate and monolithic types on the pigment separation²¹. The sample used here was pigments extracted from laves of Pleomele angustifolia, an indigenous source of natural colorants as mentioned previously. It contains common six major pigments such as chlorophylls a and b, violaxanthin, zeaxanthin, α -carotene, and β -carotene. This mestigation results showed that monolithic column provided better resolution and faster analysis, although each olumn had their characteristic features. In the present study, in addition to above three columns, an examination of two new type columns mentioned above, i.e., π -NAP and cholester columns, were conducted. This investigation approaches would give basic information to develop simple and rapid HPLC separation method for photosynthetic pigments.

2. Materials and methods

2.1. Plant material

Pleomele angustifolia Roxb. N. E. Brown was used throughout this study as a pigment source. Samples we collected from MRCPP Arboretum located in Malang, East Java, Indonesia (S 7° 57' 21.4632", E 112° 3' 24.7056"). Collected leaves were cleaned by rinsing with distilled water and were then frozen and stored at -20% for further analyses.

2.2. Columns

Chromolith[®] Performance RP-18e, 4.6 i.d. × 100 mm (MERCK, Darmstadt, Germany), Shim-Pack XR-000 3 i.d. × 100 mm (Shimadzu, Kyoto, Japan), and Shim-Pack XR-C8, 3 i.d. × 100 mm (Shimadzu) were purchase from a local provider. Cosmosil cholester, 2 i.d. × 50 mm (Nacalai Tesque), cosmosil π -NAP, 2 i.d. × 50 mm (Nacalai Tesque) were kindly gift from Nacalai Tesque, Inc., Kyoto, Japan.

2.2. Pigments extraction

P. angustifolia leaves were ground using a mortar with a few amounts of sodium ascorbate and calcular carbonate to avoid pigments oxidation and acidification. Liquid nitrogen (-196 °C) was added to prevent enzymatic reaction which can affect to the pigment stability. The homogenate (0.2 g wet weight) of *P. angustifolia* we extracted with 3 mL of 100 % methanol (GR for analysis, MERCK) in a conical bottom tube, by shaking with vortex for 10 s. In order to minimize photo-degradation and oxidation of the pigments, the extractions at measurements were carried out under green dimmed light at room temperature under ultra-high purity (99 to nitrogen atmosphere (PT. Samator, Surabaya, Indonesia). This rapid extraction method was conducted less that 1 min. Prior to injection, sample pigment was filtrated through a membrane filter (0.2 μ m, nylon, Whatman Maidstone, UK).

2.3. HPLC analysis

Pigments separation was carried out by UFLC using LC–20AD XR equipped with photodiode array detect SPD–20MA and column oven CTO–20AC (Shimadzu) as reported previously²¹. In briefly, HPLC analysis we performed isocratic method using a mobile phase consisted of acetonitrile (HPLC Grade, MERCK) and methand (GR for analysis, MERCK). The solvent ratios (v/v) were varies for analysis in the following: 20 : 80 (System I) 35 : 65 (System 2), 50 : 50 (System 3), 65 : 35 (System 4) and 80 : 20 (System 5). Column temperature used we either 30 °C or 40 °C. Pigments were detected in the range of 190 nm to 800 nm. Injection was automated by a auto–sampler SIL–20AC XR (Shimadzu) and 20 μ L pigment solution was subjected to analysis.

2.4. Pigment identification

All targeted peaks were isolated for identification. Visible absorption spectra were obtained by UV-Visible Spectrophotometer 1800 (Shimadzu) from 350 nm to 800 nm. Isolated pigments were measured in different solvents. Chlorophylls group was measured in acetone, diethyl ether, and ethanol, while carotenoids group is acetone, *n*-hexane, and ethanol. Spectral properties were then compared with those of reference spectra from the standard phytoplankton pigments^{5,6,22,23}.

2.5. Data analysis

UFLC data were reve polynomial regression was data represent an average f

3. Results and discussion

Six photosynthetic pign the properties of four coluwith comparison of absorp zeaxanthin (2nd peak), chl peak) (Table 1), as genera after separation with mobiper min and column temp Moreover, particulate pac (Fig. 1. A and C), despit column types. This is prol 40 °C, retention time is alwere also obtained by C8 of

Table 1. Identification of the pigr

Peak No.	Pigment —
1	Violaxanthin
2	Zeaxanthin
3	Chlorophyll b
4	Chlorophyll a
5	α-Carotene
6	β-Carotene

*Represent I-II-III bands for card **Mobile phase, 50 : 50 (System ** References: Hegazi⁵; Jeffrey⁶

Cholesteryl bonded sil and H) in terms of selectiv hydrophobic compounds. which have a broad spectr column could be clearly s carotene compared with C proved to be more suitab cholester column might be polar pigments (data not s This column, however, r carotenoids and their isom

Data analysis

a pigment source. Samples were (S 7° 57' 21.4632", E 112° 35' then frozen and stored at -20 °C

Germany), Shim-Pack XR-ODS, mm (Shimadzu) were purchased cosmosil π -NAP, 2 i.d. × 50 mm

f sodium ascorbate and calcium) was added to prevent enzymatic t weight) of *P. angustifolia* was cal bottom tube, by shaking with the pigments, the extractions and e under ultra-high purity (99 %) method was conducted less than filter (0.2 μ m, nylon, Whatman,

d with photodiode array detector . In briefly, HPLC analysis was C Grade, MERCK) and methanol he following: 20 : 80 (System 1); 5). Column temperature used was n. Injection was automated by an d to analysis.

ra were obtained by UV-Visible ents were measured in different anol, while carotenoids group in ose of reference spectra from the ELC data were revealed from original Shimadzu UFLC operation software, Lab Solution. Plot data and tomial regression was created by Origin 7.0 (Origin Lab Corp, Northampton, USA). Both numeric and graphic appresent an average from triplicate analyses with SE.

lesults and discussion

in photosynthetic pigments were separated with the columns used, except for π-NAP column. In here, therefore, properties of four columns were mainly compared, excluding π-NAP column. The pigments were identified icomparison of absorption spectra of isolated pigments in different solvents as follows: violaxanthin (1st peak), anthin (2nd peak), chlorophyll *b* (3rd peak), chlorophyll *a* (4th peak), α-carotene (5th peak), and β-carotene (6th (Table 1), as generally found in most of the higher plants²³⁻²⁶. Fig. 1 shows representative chromatograms reparation with mobile phase of acetonitrile-methanol, 50 : 50 (v/v) (System 3) at a fixed flow rate of 0.5 mL tim and column temperature at 30 °C and 40 °C. Rapid separation was observed in C18 than C8 column. Twee, particulate packing column needed longer time analysis than monolithic column at both temperatures (1. A and C), despite the large column volume. High column temperature enhanced time analysis in both im types. This is probably due to decrease in solvent density with increasing temperature. In both columns at *C*, retention time is able to reduce about 0.7 times of 30 °C to accomplish all peak separation. Similar results realso obtained by C8 column, XR-C8 (Fig. 1. E and F).

al. Identification of the pigments extracted from P. angustifolia

			$\lambda_{max} (nm)^*$			D 0444
Pigment	Acetone	n-Hexane	Diethyl ether	ethanol	eluent**	- Ref.***
Violaxanthin	417,440,470	416,437,469	121.44	416,438,468	413,436,465	6,22,23
Zeaxanthin	(429),450,477	(425),445,476	1 . P	(429),452,479	(420),445,472	5,6,22
Chlorophyll b	455,592,649		455,595,641	463,590,645	465,595,648	6,22,23
Chlorophyll a	430,616,662		430,616,662	430,618,666	431,617,663	5,6,22,23
α-Carotene	(423),447,475	419,443,473		421,445,473	(421),443,474	5,6,22
B -Carotene	(428),454,480	(425),449,479	-	(426),451,478	(423),450,476	5,6,22,23

resent I-II bands for carotenoids and Soret, Qx, and Qy bands for chlorophylls, parenthesis represents shoulder peak bile phase, 50 : 50 (System 3) at 40 °C eferences: Hegazi⁵; Jeffrey⁶; Britton²²; Gross²³.

the topological set of the separation of polar than hydrophobic pigments. These findings suggest that we are solution of polar than hydrophobic pigments. The separation α carotene and β -the compared with Chromolith and XR-ODS columns. On the other hand, as generally known, XR-C8 was need to be more suitable for the separation of polar than hydrophobic pigments. These findings suggest that we are compared with Chromolith and XR-ODS columns. On the other hand, as generally known, XR-C8 was need to be more suitable for the separation of polar than hydrophobic pigments. These findings suggest that we are column might be good alternative from usual C18 columns. π -NAP column was unable to separate even in a pigments (data not shown), suggesting that this column is unsuitable for separating photosynthetic pigments. It compounds, especially for the this column, however, may have advantages and potential in separating isomeric compounds, especially for menoids and their isomer separation. Further investigation is needed for optimizing this column.

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Fig. 1. UFPLC chromatograms of photosynthetic pigments from leaves of *P. angustifolia*. UFPLC was carried out an isocratic in Systa (50 : 50, v/v) and flow rate at 0.5 mL per min. Other conditions are described in the text.





To analyze time distance between pigments with different polarities, retention times of Chl_a (Chlorophyll, viol (violaxanthin), and b-car (β -carotene) were selected as peak position indicators in calculating Δt_R and t_R mathematical These pigments peaks show time distance between polar (viol) to semi-polar (Chl_a) pigments and between serpolar (Chl_a) to non-polar (b-car) pigments. Fig. 2 shows the effects of solvent compositions on Δt_R Generally reverse phase columns, separation time of pigments decreased with increasing acetonitrile concentrations (increase ionic strength). This investigation can be conventionally compared the behaviour of polar and non-polar pigmer against solvent compositions. In separation of polar pigments, Δt_R of NR-C8 column was more conspicute increased than any other columns, although other columns were almost constant. From these results, it is like concluded that under used simple mobile phase, XR-C8 has high flexible retentivity for polar pigments, indicate that this column is suitable for the separation of non-polar pigments. On the other hand, cholester column has high flexibility for non-polar pigment than any other columns. Thus this column is suitable for non-polar pigment than any other columns.

separation. The results of oprovides useful information to

Table 2	$\Delta t_{\rm R \ chl}$	_a-viol and	$\Delta t_{\rm R \ b-car-i}$
---------	--------------------------	-------------	------------------------------

No	Column
1	Chromolith
2	XR-ODS
3	XR-C8
4	Cholester

Peak retention time ratio and $t_{R\beta-car/Chl_{at}}$, were also calc column temperatures on t_R ratio XR-C8 columns, $t_{Rchl_{at/viol}}$, $t_{Rchl_{at/viol}}$, increasing solvent strength. columns were linearly decree from analyzed samples are s



Fig. 3. t_{R chl_a/viol} ratio (solid) and a (diamond) employed at 30 °

Table 3. $t_{\rm R}$ ratio polynomial relations			
No	Column		
1	Chromolith		
2	XR-ODS		
3	XR-C8		
4	Cholester		

 $\Delta t_{\rm R}$ and $t_{\rm R}$ ratio analysis pigments extracted from hacceptable results in separa for separation of polar-sem β -carotene was not the case.



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s carried out an isocratic in System 3



separation of photosynthetic pigments a are average of three experiments.

mes of Chl_a (Chlorophyll a), in calculating $\Delta t_{\rm R}$ and $t_{\rm R}$ ratio. a) pigments and between seminpositions on $\Delta t_{\rm R}$. Generally in trile concentrations (increasing polar and non-polar pigments umn was more conspicuously elester column decreased with From these results, it is likely for polar pigments, indicating and, cholester column has high uitable for non-polar pigment separation. The results of calculation by polynomial regression for columns used are shown in Table 2. This provides useful information to optimize chromatographic conditions in each column.

Table 2. $\Delta t_{\rm R \ chl}$ a-viol and $\Delta t_{\rm R \ b-car-chl}$ a polynomial regression from analyzed sample.

	0.1		T	\sim $\Delta t_{\rm R}$ chl a-viol		$\Delta t_{\rm R}$ b-car-chl a		
NO	Column		Temperature	Equation	R^2	Equation	R^2	
-	CI I'I	1001	30 °C	$Y = 4.55 - 0.06X + 0.21X^2$	0.99	$Y = 14.06 - 1.06X + 0.13X^2$	0.99	
¥.	Chromolith		40 °C	$Y = 3.54 + 0.01X + 0.10X^2$	1.00	$Y = 10.07 - 0.93X + 0.09X^2$	0.99	
2		THE OPE		30 °C	$Y = 6.66 + 0.28X + 0.23X^2$	0.94	$Y = 20.16 - 1.23X + 0.20X^2$	0.88
	XR-ODS	4	40 °C	$Y = 4.98 - 0.07X + 0.17X^2$	0.99	$Y = 14.01 - 1.18 X + 0.14 X^2$	0.99	
	110 00		30 °C	$Y = 1.66 + 0.07X + 0.06X^2$	0.99	$Y = 2.726 - 0.04X + 0.01X^2$	0.98	
3	XR-C8		40 °C	$Y = 1.36 + 0.07X + 0.03X^{2}$	0.99	$Y = 2.14 - 0.09X + 0.01X^2$	0.99	
	C1 1		30 °C	$Y = 2.91 - 0.66X + 0.15X^2$	0.99	$Y = 7.94 - 1.16X + 0.09X^2$	0.99	
4	Cholester		40 °C	$Y = 2.24 - 0.47X + 0.1X^2$	0.99	$Y = 5.37 - 0.82X + 0.06X^2$	0.99	

Peak retention time ratio (t_R ratio) is also one of parameters to understand the peak separation. Ratios of $t_{RChl_a/viol}$ and $t_{R\beta-cat/Chl_a}$, were also calculated and used as peak indicators. Fig. 3 shows the effects of solvent compositions and alumn temperatures on t_R ratio. Similar pigment separations were obtained in both temperatures. In XR-ODS and XR-C8 columns, $t_{Rchl_a/viol}$ was almost constant up to solvent composition of 50 : 50, but then increased with increasing solvent strength. This tendency was also observed in Δt_R . On the other hand, $t_{R\beta-cat/chl_a}$ calculated from all alumns were linearly decreased with increasing solvent strength, but their values were low. Polynomial regression from analyzed samples are summarized in Table 3.



(diamond) employed at 30 °C and 40 °C column temperature.
(diamond) employed at 30 °C and 40 °C column temperature.

Table 3. t_R ratio polynomial regression calculated from analyzed sample.

No	Column	Temperature	IR chl a/viol		IR b-car/chl a	
			Equation	\mathbb{R}^2	Equation	R^2
1	Chromolith	30 °C	$Y = 10.14 + 0.20X + 0.02X^2$	0.96	$Y = 4.00 - 0.48X + 0.02X^2$	0.99
		40 °C	$Y = 10.54 - 0.41X + 0.07X^2$	0.72	$Y = 3.68 - 0.42X + 0.02X^2$	0.99
	XR-ODS	30 °C	$Y = 13.05 - 0.40X + 0.24X^2$	0.96	$Y = 4.05 - 0.45X + 0.02X^2$	0.99
2		40 °C	$Y = 12.68 - 0.68X + 0.23X^2$	0.98	$Y = 3.72 - 0.39X + 0.01X^2$	0.99
3	XR-C8	30 °C	$Y = 8.04 - 0.95X + 0.28X^2$	0.95	$Y = 2.47 - 0.20X + 0.01X^2$	0.99
		40 °C	$Y = 7.16 - 0.36X + 0.13X^2$	0.98	$Y = 2.39 - 0.20X + 0.01X^2$	0.99
4	Cholester	30 °C	$Y = 4.85 - 0.58X + 0.14X^2$	0.99	$Y = 3.40 - 0.14X - 0.02X^2$	0.99
		40 °C	$Y = 4.36 - 0.54X + 0.12X^2$	0.99	$Y = 3.00 - 0.13X - 0.02X^2$	0.99

 Δt_R and t_R ratio analysis had provided clear description for the column performance in separating photosynthetic rements extracted from *P. angustifolia*. All investigated columns, except cosmosil π -NAP column, provide acceptable results in separating pigments from polar to non-polar species. Most of these columns had their abilities for separation of polar-semi polar pigments. However, separation of non-polar carotenoids such as α -carotene and l-carotene was not the case.

Subsequently, this investigation conducted Gaussian peak fitting analysis using Origin software to determine resolution of columns. This analysis focused on the peaks of structurally similar pigments, α -carotene and carotene (Fig. 4). Under used conditions, poor pigment separation was observed in the XR-C8. Similarly Chromote column gave low resolution probably due to peak broadening. XR-ODS provided good results of the separation to much high resolution was obtained by cholester column. Combined together with the previous results, cholest column is superior for the separation of non-polar pigments in terms of selectivity and resolution.



Fig. 4. Peak separation of non-polar pigments (α-carotene and β-carotene). UFPLC was carried out an isocratic in Sys. 3 and 0.5 mL perminfor rate. Black line represents original chromatogram. Red lines show Gaussian peak fitting results.

Previously, particulate packing columns (C8 and C18) had been commonly used in HPLC for separation photosynthetic pigments. Huge effort has been made by many researchers to optimize pigment separation throug these column types^{5,10,14,15,27}. Most of them employed gradient method as a strategy to increase in separation quark. In some HPLC methods, a narcotic and psychotropic source material, acetone, is used as mobile phase^{5,14,15}. Since the adoption of the 1988 UN Convention against Illicit Traffic in Narcotic Drugs and Psychotropic Substances, is some countries including Indonesia, those solvents trading right was limited under very close supervision in ordert minimize irresponsible used²⁸. This restriction was giving us new issue in providing better method for HPLC, while is not use of drug related solvents. Low time and solvent consuming analysis was also becoming strong demand in pigment separation analysis due to environmental problem and stability. Photosynthetic pigments were unstable against extreme uncontrolled environment. Long time HPLC analysis should be considered solvent-pigme interaction and column temperature which gives effect in pigment stability^{29,30}. This may cause in decreasing accuracy of the data.

In the previous study²¹, the efficiency between particulate packing and monolithic columns were compare Clearly different from particulate packing bed, monolith column composed by a continuous character of skelete which fulfills the separation chambers. Monolith contained a discrete bimodal pore size distribution^{13,31}. Chromolic column showed a typical characteristic of monolithic column in the separation of *P. angustifolia* pigments is provided better resolution and faster analysis. Thus, high tolerates to flow rate system of this column provides us optimize a rapid separation method.

Cosmosil cholester column is claimed as their abilities of enhanced selectivity over traditional C18 materials a greater performance in separating isomers or other closely related compounds. It is expected as an ideal column of method development and serves as an excellent alternative to traditional C18 columns. There was, however, limit information about this column performance relating to photosynthetic pigment separation. In this report, this colum has shown its performance compared to other columns. This column has proved its advantages and specialize characteristic in separating hydrophobic pigment in such a rapid elution time. This is the first report on the separation of photosynthetic pigment by cosmosil cholester column.

4. Conclusion

In this study, the efficien separation of photosynthetic Among above columns tester especially for hydrophobic p superior to resolution of struc parameters in separating phot

Acknowledgement

This project was supported 2012 and RT–2014–0432, No lptek) Program (SK No. 284, authors also special thanks to columns.

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4. Conclusion

In this study, the efficiency of five reverse-phase columns, C8, C18, C18 monolithic, π -NAP, and cholester, for separation of photosynthetic pigments at several fixed conditions of mobile phase and temperature were compared. Among above columns tested, cholester column is suitable for separation of pigments for a broad range of polarity, specially for hydrophobic pigments in rapid elution time and simple mobile phase. In addition, this column is also superior to resolution of structurally similar pigments. These findings can help in the selection of column and HPLC parameters in separating photosynthetic pigments by using simple mobile phase system.

Acknowledgement

This project was supported by National Innovation System Research Grant (RT–2013–0172, No: 187/M/Kp/XI/ 2012 and RT–2014–0432, No: 288/M/Kp/XII/2013) and National Research Center of Excellence (Pusat Unggulan (ptck) Program (SK No. 284/M/Kp/XI/2013) provided by Indonesian Ministry of Research and Technology. The authors also special thanks to Nacalai Tesque, Inc. (Kyoto, Japan) for their gift of cosmosil cholester and π -NAP columns.

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R-C8. Similarly Chromolith

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the pigment separation through increase in separation quality. If as mobile phase^{5,14,15}. Since d Psychotropic Substances, in ty close supervision in order to etter method for HPLC, which the becoming strong demand for thetic pigments were unstable e considered solvent-pigment this may cause in decreasing

hic columns were compared. ntinuous character of skeleton, ze distribution^{13,31}. Chromolith f *P. angustifolia* pigments. It n of this column provides us to

r traditional C18 materials and xpected as an ideal column for s. There was, however, limited tion. In this report, this column its advantages and specialized his is the first report on the

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Abstract

Spirulina is one of th Bioactive compound in Sp problem in the world. The microalgae was conducted antihyperglycemic activity phycocyanin. Blood glucos administration of biomass a blood glucose level.

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* Corresponding author. Tel.: + E-mail address: iriani25@gma

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Procedia Chemistry 14 (2015) 202 - 210

2nd Humboldt Kolleg in conjunction with International Conference on Natural Sciences, HK-ICONS 2014

Separation of Photosynthetic Pigments by High-Performance Liquid Chromatography: Comparison of Column Performance, Mobile Phase, and Temperature

Indriatmoko^a, Yuzo Shioi^a, Tatas Hardo Panintingjati Brotosudarmo^a, Leenawaty Limantara^a*

^aMa Chung Research Center for Photosynthetic Pigments, Universitas Ma Chung, Malang 65151, East Java, Indonesia

Abstract

High-performance liquid chromatography (HPLC) has been commonly used as method of separating and identifying photosynthetic pigments such as chlorophylls and carotenoids because of such advantages as speed, high resolution and sensitivity. In this technique, high separation relies largely on the type of column material. This study compared the efficiency of five reverse-phase columns, C8, C18, C18 monolithic, π -NAP, and cholester, for separation of photosynthetic pigments at several fixed conditions of mobile phase and temperature. This investigation also analysed the parameters of Δt_R and t_R ratio for selected pigments and resolution for structural isomers, such as α - and β -carotene. Among above columns tested, cholester column is suitable for separation of pigments not only for a broad range of polarity, but also for hydrophobic pigments in a simple mobile phase. This finding can help in the selection of column and HPLC parameters in separating photosynthetic pigments.

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Keyword: Cholesteryl bonded; HPLC column; monolithic packing; particulate packing; photosynthetic pigments; reverse phase.

*Corresponding author. Tel: +62 813 2636 0303, Fax+62 341 550 175 *E-mail address:* leenawaty.limantara@machung.ac.id

Nomenclature					
$\Delta t_{\rm R}$	retention time difference				
t _R	retention time				
t _R Ratio	ratio between retention time of two pigment peaks				
v/v	solvents volume ratio				

1. Introduction

Chromatography method has been introduced since 1905 as specialized technique for photosynthetic pigments separation¹. Since then, several methods have been developed and commonly used, e.g., thin-layer chromatography^{1,2}, column chromatography^{3,4}, and high-performance liquid chromatography (HPLC)^{5–7}. Ultra-fast liquid chromatography (UFLC) was one of the newest generations of HPLC which provide special advantages in conducting researches with low time consuming and high resolution data^{8,9}. These advantages gave an opportunity in the development of a low cost and rapid analysis method. Generally, quality of UFLC separation is affected by four main parameters, i.e., mobile phase, flow rate, column temperature, and column type. Thus, an exploration of the best conditions for pigment separation had become a challenge for chromatography researchers in the world. Pigment analyses of leaves of higher plants were reported using different UFLC/HPLC analytical methods^{10–12}.

In main separation parameters, column material has been understood as an important part where pigment separation occurs during analysis. Other parameters are usually set depend on column type. Generally, HPLC/UFLC columns are distinguished as monolithic and particulate packing types¹³. Ability of these columns in separating pigments had reported for various samples and their improvements^{5,14–16}. Monolithic column is consisted of small-sized skeletons and wide through-pores which can be achieved higher separation efficiency than the case with particulate packing columns at a similar pressure drop¹⁷. There are several reports on the monolithic column performance in separating photosynthetic pigments^{16,18,19}. This column type is known for its advantages in providing good separation and short time analysis²⁰. Particulate packing columns have also been widely used for pigment separation. Two most well-used particulate packing column are octyl (C8) and octadecyl (C18) types based on silica. There are numbers of reports on these C8 and C18 which used to develop optimized method for analysis of photosynthetic pigments^{5,14,15}.

Recently, new types of column based on napthylethtyl bonded silica packing and cholesteryl bonded silica packing were invented. Cholester column is basically similar with conventional ODS column as their equivalent hydrophobicity. Nevertheless, cholester column has high sensitivity for hydrophobic compound due to their strong stereo-selectivity. Like cholester column, π -NAP column has unique specific selectivity in separation. This napthylethyl bonded silica packing column was built for π - π interactions for hydrophobic compound. These advantages, in case of photosynthetic pigments separation, provide better chance to provide good separation of carotenes group.

In previous investigation, two silica particulate packing columns (C18 and C8) were analysed as the standard in addition to C18 monolithic type column to understand the effect of carbon chain length and the difference between particulate and monolithic types on the pigment separation²¹. The sample used here was pigments extracted from leaves of *Pleomele angustifolia*, an indigenous source of natural colorants as mentioned previously. It contains common six major pigments such as chlorophylls *a* and *b*, violaxanthin, zeaxanthin, α -carotene, and β -carotene. This investigation results showed that monolithic column provided better resolution and faster analysis, although each column had their characteristic features. In the present study, in addition to above three columns, an examination of two new type columns mentioned above, i.e., π -NAP and cholester columns, were conducted. This investigation approaches would give basic information to develop simple and rapid HPLC separation method for photosynthetic pigments.

2. Materials and methods

2.1. Plant material

Pleomele angustifolia Roxb. N. E. Brown was used throughout this study as a pigment source. Samples were collected from MRCPP Arboretum located in Malang, East Java, Indonesia (S 7° 57' 21.4632", E 112° 35' 24.7056"). Collected leaves were cleaned by rinsing with distilled water and were then frozen and stored at -20 °C for further analyses.

2.2. Columns

Chromolith[®] Performance RP-18e, 4.6 i.d. × 100 mm (MERCK, Darmstadt, Germany), Shim-Pack XR-ODS, 3 i.d. × 100 mm (Shimadzu, Kyoto, Japan), and Shim-Pack XR-C8, 3 i.d. × 100 mm (Shimadzu) were purchased from a local provider. Cosmosil cholester, 2 i.d. × 50 mm (Nacalai Tesque), cosmosil π -NAP, 2 i.d. × 50 mm (Nacalai Tesque) were kindly gift from Nacalai Tesque, Inc., Kyoto, Japan.

2.2. Pigments extraction

P. angustifolia leaves were ground using a mortar with a few amounts of sodium ascorbate and calcium carbonate to avoid pigments oxidation and acidification. Liquid nitrogen (-196 °C) was added to prevent enzymatic reaction which can affect to the pigment stability. The homogenate (0.2 g wet weight) of *P. angustifolia* was extracted with 3 mL of 100 % methanol (GR for analysis, MERCK) in a conical bottom tube, by shaking with vortex for 10 s. In order to minimize photo-degradation and oxidation of the pigments, the extractions and measurements were carried out under green dimmed light at room temperature under ultra-high purity (99 %) nitrogen atmosphere (PT. Samator, Surabaya, Indonesia). This rapid extraction method was conducted less than 1 min. Prior to injection, sample pigment was filtrated through a membrane filter (0.2 μ m, nylon, Whatman, Maidstone, UK).

2.3. HPLC analysis

Pigments separation was carried out by UFLC using LC–20AD XR equipped with photodiode array detector SPD–20MA and column oven CTO–20AC (Shimadzu) as reported previously²¹. In briefly, HPLC analysis was performed isocratic method using a mobile phase consisted of acetonitrile (HPLC Grade, MERCK) and methanol (GR for analysis, MERCK). The solvent ratios (v/v) were varies for analysis in the following: 20 : 80 (System 1); 35 : 65 (System 2), 50 : 50 (System 3), 65 : 35 (System 4) and 80 : 20 (System 5). Column temperature used was either 30 °C or 40 °C. Pigments were detected in the range of 190 nm to 800 nm. Injection was automated by an auto–sampler SIL–20AC XR (Shimadzu) and 20 µL pigment solution was subjected to analysis.

2.4. Pigment identification

All targeted peaks were isolated for identification. Visible absorption spectra were obtained by UV-Visible Spectrophotometer 1800 (Shimadzu) from 350 nm to 800 nm. Isolated pigments were measured in different solvents. Chlorophylls group was measured in acetone, diethyl ether, and ethanol, while carotenoids group in acetone, *n*-hexane, and ethanol. Spectral properties were then compared with those of reference spectra from the standard phytoplankton pigments^{5,6,22,23}.

2.5. Data analysis

UFLC data were revealed from original Shimadzu UFLC operation software, Lab Solution. Plot data and polynomial regression was created by Origin 7.0 (Origin Lab Corp, Northampton, USA). Both numeric and graphic data represent an average from triplicate analyses with SE.

3. Results and discussion

Six photosynthetic pigments were separated with the columns used, except for π -NAP column. In here, therefore, the properties of four columns were mainly compared, excluding π -NAP column. The pigments were identified with comparison of absorption spectra of isolated pigments in different solvents as follows: violaxanthin (1st peak), zeaxanthin (2nd peak), chlorophyll *b* (3rd peak), chlorophyll *a* (4th peak), α -carotene (5th peak), and β -carotene (6th peak) (Table 1), as generally found in most of the higher plants^{23–26}. Fig. 1 shows representative chromatograms after separation with mobile phase of acetonitrile-methanol, 50 : 50 (v/v) (System 3) at a fixed flow rate of 0.5 mL per min and column temperature at 30 °C and 40 °C. Rapid separation was observed in C18 than C8 column. Moreover, particulate packing column needed longer time analysis than monolithic column at both temperatures (Fig. 1. A and C), despite the large column volume. High column temperature enhanced time analysis in both column types. This is probably due to decrease in solvent density with increasing temperature. In both columns at 40 °C, retention time is able to reduce about 0.7 times of 30 °C to accomplish all peak separation. Similar results were also obtained by C8 column, XR-C8 (Fig. 1. E and F).

Table 1. Identification of the pigments extracted from P. angustifolia

Peak	Pigment	$\lambda_{\max} (nm)^*$					
No.		Acetone	n-Hexane	Diethyl ether	ethanol	eluent**	Kel.***
1	Violaxanthin	417,440,470	416,437,469	-	416,438,468	413,436,465	6,22,23
2	Zeaxanthin	(429),450,477	(425),445,476	-	(429),452,479	(420),445,472	5,6,22
3	Chlorophyll b	455,592,649	-	455,595,641	463,590,645	465,595,648	6,22,23
4	Chlorophyll a	430,616,662	-	430,616,662	430,618,666	431,617,663	5,6,22,23
5	α-Carotene	(423),447,475	419,443,473	-	421,445,473	(421),443,474	5,6,22
6	β-Carotene	(428),454,480	(425),449,479	-	(426),451,478	(423),450,476	5,6,22,23

*Represent I-II-III bands for carotenoids and Soret, Qx, and Qy bands for chlorophylls, parenthesis represents shoulder peak

**Mobile phase, 50 : 50 (System 3) at 40 °C

***References: Hegazi⁵; Jeffrey⁶; Britton²²; Gross²³.

Cholesteryl bonded silica packing column was superior for separation among all columns examined (Fig. 1.G and H) in terms of selectivity and resolution of hydrophobic pigments, as suggested by manufacturer for separating hydrophobic compounds. This investigation examined suitability for the separation of photosynthetic pigments which have a broad spectrum of polarity. As shown in Fig. 1.G and H (see peaks 5 and 6) and also Fig. 4, cholester column could be clearly separated not only polar pigments, but also non-polar pigments, *trans* α -carotene and β -carotene compared with Chromolith and XR-ODS columns. On the other hand, as generally known, XR-C8 was proved to be more suitable for the separation of polar than hydrophobic pigments. These findings suggest that cholester column might be good alternative from usual C18 columns. π -NAP column was unable to separate even in polar pigments (data not shown), suggesting that this column is unsuitable for separating photosynthetic pigments. This column, however, may have advantages and potential in separating isomeric compounds, especially for carotenoids and their isomer separation. Further investigation is needed for optimizing this column.



Fig. 1. UFPLC chromatograms of photosynthetic pigments from leaves of *P. angustifolia*. UFPLC was carried out an isocratic in System 3 (50:50, v/v) and flow rate at 0.5 mL per min. Other conditions are described in the text.



Fig. 2. $\Delta t_{\text{Rchl},a-\text{viol}}$ (solid circle) and $\Delta t_{\text{Rb-car-chl},a}$ (open circle) were calculated from the results of UFPLC separation of photosynthetic pigments extracted from leaves of *P. angustifolia*. Other conditions are the same as in Fig. 1. Data are average of three experiments. SE is less than ± 0.5 .

To analyze time distance between pigments with different polarities, retention times of Chl_a (Chlorophyll a), viol (violaxanthin), and b-car (β -carotene) were selected as peak position indicators in calculating Δt_R and t_R ratio. These pigments peaks show time distance between polar (viol) to semi-polar (Chl_a) pigments and between semi-polar (Chl_a) to non-polar (b-car) pigments. Fig. 2 shows the effects of solvent compositions on Δt_R . Generally in reverse phase columns, separation time of pigments decreased with increasing acetonitrile concentrations (increasing ionic strength). This investigation can be conventionally compared the behaviour of polar and non-polar pigments against solvent compositions. In separation of polar pigments, Δt_R of XR-C8 column was more conspicuously increased than any other columns. In contrast, Δt_R of non-polar pigments in cholester column decreased with increasing acetonitrile concentrations, although other columns were almost constant. From these results, it is likely concluded that under used simple mobile phase, XR-C8 has high flexible retentivity for polar pigments, indicating that this column is suitable for the separation of non-polar pigments. On the other hand, cholester column has high flexibility for non-polar pigment than any other columns. Thus this column is suitable for non-polar pigment.

separation. The results of calculation by polynomial regression for columns used are shown in Table 2. This provides useful information to optimize chromatographic conditions in each column.

No	Column	Temperature	$\Delta t_{\rm R \ chl}$ a-viol		$\Delta t_{\rm R \ b-car-chl \ a}$	
			Equation	\mathbb{R}^2	Equation	R ²
1	Chromolith	30 °C	$Y = 4.55 - 0.06X + 0.21X^2$	0.99	$Y = 14.06 - 1.06X + 0.13X^2$	0.99
I		40 °C	$Y = 3.54 + 0.01X + 0.10X^2$	1.00	$Y = 10.07 - 0.93X + 0.09X^2$	0.99
2	XR-ODS	30 °C	$Y = 6.66 + 0.28X + 0.23X^2$	0.94	$Y = 20.16 - 1.23X + 0.20X^2$	0.88
2		40 °C	$Y = 4.98 - 0.07X + 0.17X^2$	0.99	$Y = 14.01 - 1.18 X + 0.14 X^2$	0.99
2	XR-C8	30 °C	$Y = 1.66 + 0.07X + 0.06X^2$	0.99	$Y = 2.726 - 0.04X + 0.01X^2$	0.98
3		40 °C	$Y = 1.36 + 0.07X + 0.03X^2$	0.99	$Y = 2.14 - 0.09X + 0.01X^2$	0.99
4	Cholester	30 °C	$Y = 2.91 - 0.66X + 0.15X^2$	0.99	$Y = 7.94 - 1.16X + 0.09X^2$	0.99
4		40 °C	$Y = 2.24 - 0.47X + 0.1X^2$	0.99	$Y = 5.37 - 0.82X + 0.06X^2$	0.99

Table 2. $\Delta t_{R \text{ chl}_a\text{-viol}}$ and $\Delta t_{R \text{ b-car-chl}_a}$, polynomial regression from analyzed sample.

Peak retention time ratio (t_R ratio) is also one of parameters to understand the peak separation. Ratios of $t_{RChl_a/viol}$ and $t_{R\beta-car/Chl_a}$, were also calculated and used as peak indicators. Fig. 3 shows the effects of solvent compositions and column temperatures on t_R ratio. Similar pigment separations were obtained in both temperatures. In XR-ODS and XR-C8 columns, $t_{Rchl_a/viol}$ was almost constant up to solvent composition of 50 : 50, but then increased with increasing solvent strength. This tendency was also observed in Δt_R . On the other hand, $t_{R\beta-car/chl_a}$ calculated from all columns were linearly decreased with increasing solvent strength, but their values were low. Polynomial regression from analyzed samples are summarized in Table 3.



Fig. 3. *t*_R chl_a/viol ratio (solid) and *t*_R b-car/chl_a</sub> ratio (open), from Chromolith (Square), XR-ODS (triangle), XR-C8 (circle), and cholester column (diamond) employed at 30 °C and 40 °C column temperature.

No	Column	Tommonotuno	t _{R chl a/viol}		$t_{\rm R\ b-car/chl\ a}$	
		remperature	Equation	R ²	Equation	\mathbb{R}^2
1	Chromolith	30 °C	$Y = 10.14 + 0.20X + 0.02X^2$	0.96	$Y = 4.00 - 0.48X + 0.02X^2$	0.99
		40 °C	$Y = 10.54 - 0.41X + 0.07X^2$	0.72	$Y = 3.68 - 0.42X + 0.02X^2$	0.99
2	XR-ODS	30 °C	$Y = 13.05 - 0.40X + 0.24X^2$	0.96	$Y = 4.05 - 0.45X + 0.02X^2$	0.99
		40 °C	$Y = 12.68 - 0.68X + 0.23X^2$	0.98	$Y = 3.72 - 0.39X + 0.01X^2$	0.99
3	XR-C8	30 °C	$Y = 8.04 - 0.95X + 0.28X^2$	0.95	$Y = 2.47 - 0.20X + 0.01X^2$	0.99
		40 °C	$Y = 7.16 - 0.36X + 0.13X^2$	0.98	$Y = 2.39 - 0.20X + 0.01X^2$	0.99
4	Cholester	30 °C	$Y = 4.85 - 0.58X + 0.14X^2$	0.99	$Y = 3.40 - 0.14X - 0.02X^2$	0.99
		40 °C	$Y = 4.36 - 0.54X + 0.12X^2$	0.99	$Y = 3.00 - 0.13X - 0.02X^2$	0.99

Table 3. t_R ratio polynomial regression calculated from analyzed sample.

 $\Delta t_{\rm R}$ and $t_{\rm R}$ ratio analysis had provided clear description for the column performance in separating photosynthetic pigments extracted from *P. angustifolia*. All investigated columns, except cosmosil π -NAP column, provide acceptable results in separating pigments from polar to non-polar species. Most of these columns had their abilities for separation of polar-semi polar pigments. However, separation of non-polar carotenoids such as α -carotene and β -carotene was not the case.

Subsequently, this investigation conducted Gaussian peak fitting analysis using Origin software to determine the resolution of columns. This analysis focused on the peaks of structurally similar pigments, α -carotene and β -carotene (Fig. 4). Under used conditions, poor pigment separation was observed in the XR-C8. Similarly Chromolith column gave low resolution probably due to peak broadening. XR-ODS provided good results of the separation, but much high resolution was obtained by cholester column. Combined together with the previous results, cholester column is superior for the separation of non-polar pigments in terms of selectivity and resolution.



Fig. 4. Peak separation of non-polar pigments (α-carotene and β-carotene). UFPLC was carried out an isocratic in Sys. 3 and 0.5 mL per min flow rate. Black line represents original chromatogram. Red lines show Gaussian peak fitting results.

Previously, particulate packing columns (C8 and C18) had been commonly used in HPLC for separating photosynthetic pigments. Huge effort has been made by many researchers to optimize pigment separation through these column types^{5,10,14,15,27}. Most of them employed gradient method as a strategy to increase in separation quality. In some HPLC methods, a narcotic and psychotropic source material, acetone, is used as mobile phase^{5,14,15}. Since the adoption of the 1988 UN Convention against Illicit Traffic in Narcotic Drugs and Psychotropic Substances, in some countries including Indonesia, those solvents trading right was limited under very close supervision in order to minimize irresponsible used²⁸. This restriction was giving us new issue in providing better method for HPLC, which is not use of drug related solvents. Low time and solvent consuming analysis was also becoming strong demand for pigment separation analysis due to environmental problem and stability. Photosynthetic pigments were unstable against extreme uncontrolled environment. Long time HPLC analysis should be considered solvent-pigment interaction and column temperature which gives effect in pigment stability^{29,30}. This may cause in decreasing accuracy of the data.

In the previous study²¹, the efficiency between particulate packing and monolithic columns were compared. Clearly different from particulate packing bed, monolith column composed by a continuous character of skeleton, which fulfills the separation chambers. Monolith contained a discrete bimodal pore size distribution^{13,31}. Chromolith column showed a typical characteristic of monolithic column in the separation of *P. angustifolia* pigments. It provided better resolution and faster analysis. Thus, high tolerates to flow rate system of this column provides us to optimize a rapid separation method.

Cosmosil cholester column is claimed as their abilities of enhanced selectivity over traditional C18 materials and greater performance in separating isomers or other closely related compounds. It is expected as an ideal column for method development and serves as an excellent alternative to traditional C18 columns. There was, however, limited information about this column performance relating to photosynthetic pigment separation. In this report, this column has shown its performance compared to other columns. This column has proved its advantages and specialized characteristic in separating hydrophobic pigment in such a rapid elution time. This is the first report on the separation of photosynthetic pigment by cosmosil cholester column.

4. Conclusion

In this study, the efficiency of five reverse-phase columns, C8, C18, C18 monolithic, π -NAP, and cholester, for separation of photosynthetic pigments at several fixed conditions of mobile phase and temperature were compared. Among above columns tested, cholester column is suitable for separation of pigments for a broad range of polarity, especially for hydrophobic pigments in rapid elution time and simple mobile phase. In addition, this column is also superior to resolution of structurally similar pigments. These findings can help in the selection of column and HPLC parameters in separating photosynthetic pigments by using simple mobile phase system.

Acknowledgement

This project was supported by National Innovation System Research Grant (RT–2013–0172, No: 187/M/Kp/ XI/ 2012 and RT–2014–0432, No: 288/M/Kp/XII/2013) and National Research Center of Excellence (Pusat Unggulan Iptek) Program (SK No. 284/M/Kp/XI/2013) provided by Indonesian Ministry of Research and Technology. The authors also special thanks to Nacalai Tesque, Inc. (Kyoto, Japan) for their gift of cosmosil cholester and π -NAP columns.

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