



# CERTIFICATE of Appreciation

This is to certify that

**Dr. Hendrik Oktendy Lintang**

contributed as

**Oral Presenter**

in Ma Chung International Conference on Chromatography  
9 - 11 October 2017, Malang, Indonesia

**Organized by:**



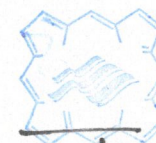
Ma Chung Research Center  
for Photosynthetic Pigments



*[Signature]*

**Dr. Leny Yulianti**

Organizing Committee Chair  
MIC-Chroma



Ma Chung Research Center  
for Photosynthetic Pigments

*[Signature]*

**Tatas H.P. Brotosudarmo, Ph.D.**

Director of Ma Chung Research Center  
for Photosynthetic Pigments

**Supported by:**







## LETTER OF ASSIGNMENT

No: 309B/MACHUNG/ST/X/2017

The Rector of Universitas Ma Chung hereby assigns:

Name : Dr.Eng. Hendrik Oktendy Lintang, S.Si., M.Eng.  
Employee's Number : 20160017  
Position : Principal Investigator of MRCPP.

to participate in the **Ma Chung International Conference on Chromatography (MIC-Chroma) as Oral Presenter** with abstract entitled "**Porous Kaolin-Phosphotungstic Acid Composites as Heterogeneous Catalyst for Friedel-Crafts Acylation of Anisole**" hosted by MRCPP Universitas Ma Chung which is held on 9-11 Oktober 2017 in Ijen Suites Resort and Convention, Malang.

He has to submit an official report when returns to work.

Please be informed.

Malang, 9 October 2017

Rector



Dr. Chatief Kunjaya

Acknowledged by,



(Name)

CC:

1. Vice Rector
2. Ma Chung Research Center for Photosynthetic Pigments (MRCPP)
3. Human Resource Management



UNIVERSITAS  
**MA CHUNG**



Ma Chung Research Center  
for Photosynthetic Pigments

# MIC-Chroma

Ma Chung International Conference on Chromatography

## Abstracts & Program

MIC-Chroma

9 - 11 October 2017  
Malang, Indonesia

In collaboration with :



With the support of:





## GENERAL SCHEDULE

Tuesday, 10 October 2017

| Time          | Ballroom   | Classroom  |
|---------------|--|--|
| 08:00 - 09:15 | Registration   |  |
| 09:15 - 09:25 | Opening remark   |  |
| 09:25 - 10:15 | Plenary lecture (PL-01):<br><b>Prof. Dr. Hian Kee Lee</b><br><i>"Miniaturized and Green Sample Preparation Procedures in Combination with Chromatographic Analysis"</i><br><br>Moderator: Tatas H.P. Brotosudarmo, Ph.D. |  |
| 10:15 - 10:30 | Coffee break   |  |
| 10:30 - 11:05 | Keynote lecture (KL-01):<br><b>Prof. Dr. Mohd Marsin Sanagi</b><br><i>"Recent Advances in Solid Phase Microextraction Techniques towards Green Chemical Analysis"</i><br><br>Moderator: Dr. Leny Yuliati                 |  |
| 11:05 - 11:40 | Keynote lecture (KL-02):<br><b>Assoc. Prof. Dr. Koichiro Awai, Ph.D.</b><br><i>"Chromatography on Lipid Analysis: From Traditional Methods to Advanced Technologies"</i><br><br>Moderator: Dr. Hendrik Oktendy Lintang   |  |
| 11:40 - 12:00 | Group photo  |  |
| 12:00 - 13:00 | Lunch break  |  |
| 13:00 - 14:30 | Poster session   |  |
| 14:30 - 14:55 | Invited speaker (IS-01):<br><b>Dr. Zhaoqi Zhan</b><br><i>"The Shimadzu MRM Method Package Tool For Rapid Method Development on LC/MS/MS for Various Applications"</i><br><br>Moderator: Dr. Hendrik Oktendy Lintang      | Invited speaker (IS-02):<br><b>Wangsa Tirta Ismaya, Ph.D.</b><br><i>"Development of HPLC Analysis of Protein Based Therapeutic Drug Product"</i><br><br>Moderator: Rehmadata Sitepu, S.Farm, M.Si., Apt. |



|               |  |   |
|---------------|--|---|
| 14:55 - 15:20 | <p>Invited speaker (IS-03):<br/><b>Dr. rer.nat Rino R. Mukti</b><br/><i>"Recent Advances in the Syntheses of Zeolites and Their Emerging Applications"</i></p> <p>Moderator: Dr. Yuyun Yuniati</p> | <p>Invited speaker (IS-04):<br/><b>Tatas H.P. Brotosudarmo,</b><br/><b>Dipl.Chem., Ph.D., MRSC.</b><br/><i>"Analysis of Major Photosynthetic Pigment from Marine Brown Alga"</i></p> <p>Moderator: Monika Prihastyanti,<br/>M.Nat.Sc.</p> |
| 15:20 - 15:35 | Coffee break   |   |
| 15:35 - 16:35 | <p>Oral presentation</p> <p>Moderator:<br/>1. Dr. Hendrik Oktendy Lintang<br/>2. Dr. Yuyun Yuniati</p>   | <p>Oral presentation</p> <p>Moderator:<br/>1. Rehmadata Sitepu, S.Farm.,<br/>M.Si., Apt.<br/>2. Monika Prihastyanti, M.Nat.Sc.</p>  |
| 18:00         | Gala dinner  |   |

**Wednesday, 11 October 2017**

| Time          | Ballroom   | Classroom   |
|---------------|--|---|
| 08:00 - 09:00 | Registration   |   |
| 09:00 - 09:25 | <p>Invited speaker (IS-05):<br/><b>Prof. Dr.rer.nat. Gunawan Indrayanto</b><br/><i>"Validation of Chromatographic Methods of Analysis: Application for Herbal Drugs"</i></p> <p>Moderator: Dion Notario,<br/>S.Farm., M.Sc., Apt.</p>                      | <p>Invited speaker (IS-06):<br/><b>Prof. Dr. Suppa Hannongbua</b><br/><i>"Determination of Absolute Configuration of Natural Products by Computational Approaches"</i></p> <p>Moderator: Roki Alfanaar, S.Si.,<br/>M.Si.</p>  |
| 09:25 - 09:50 | <p>Invited speaker (IS-07):<br/><b>Assoc. Prof. Dr. Chua Lee Suan</b><br/><i>"LC-MS/MS Based Phytochemical Profiling Integrated with Chemometrics for Quality Assurance of Herbal Plants"</i></p> <p>Moderator: Ruth Febriana<br/>Kesuma, S.Si., M.Si.</p> | <p>Invited speaker (IS-08):<br/><b>Assoc. Prof. Dr.Sc. Akhmad Sabaruddin, S.Si., M.Sc.</b><br/><i>"Development of Organic Polymer-based Monoliths: Application to Analytical and Bioanalytical Chemistry by Liquid Chromatography"</i></p> <p>Moderator: Eva Monica, S.Farm.,<br/>M.Sc., Apt.</p> |
| 09:50 - 10:05 | Coffee break   |   |



|               |   |  |
|---------------|---|--|
| 10:05 - 11:50 | Oral presentation<br><br>Moderator:<br>1. Dion Notario, S.Farm, M.Sc.,<br>Apt.<br>2. Ruth Febriana Kesuma, S.Si.,<br>M.Si.  | Oral presentation<br><br>Moderator:<br>1. Rokiy Alfanaar, S.Si., M.Si.<br>2. Eva Monica, S.Farm., M.Sc., Apt.        |
| 11:50 - 13:00 | Lunch break   |  |
| 13:00 - 14:15 | Oral presentation<br><br>Moderator:<br>1. Heriyanto, S.Si., M.Si., M.Sc.<br>2. Martanty Aditya, M.Farm-<br>Klin., Apt.      | Oral presentation<br><br>Moderator:<br>1. Rollando, S.Farm., M.Sc., Apt.<br>2. Renny Indrawati, .S.TP.,<br>M.Nat.Sc. |
| 14:15 - 14:30 | Coffee break  |  |
| 14:30 - 16:30 | ABP session :<br><b>Prof. Dr. Reiko Motohashi</b><br><b>Dr. Shin Usuki</b><br><br>Moderator: Tatas H.P. Brotosudarmo, Ph.D. |  |
| 16:30         | Closing ceremony  |  |



## SCHEDULE OF ORAL PRESENTATIONS

Tuesday, 10 October 2017

| Time             | Ballroom  | Classroom  |
|------------------|---|--|
| 15.35 – 15.50    | <b>Faisal Hussin (OP-01)</b><br><i>“Highly Efficient Zinc Oxide-Carbon Nitride Hybrid Photocatalysts for Degradation of Phenol under UV and Visible Light Irradiation”</i>  | <b>Siti Maryam Jasman (OP-05)</b><br><i>“Photocatalytic Oxidation of Nitrite Ion over Carbon Nitride”</i>  |
| 15:50 – 16:05    | <b>Bactiar R.P. Ihsan (OP-02)</b><br><i>“Validation Method HPLC for analysis Andrographolide in Ethyl Acetate Fractions of 70% Ethanol Extracts Andrographis paniculata”</i>  | <b>Darius Greenidge (OP-06)</b><br><i>“Investigations of Color Center Phenomena in Cinnabar and Metacinnabar through Electron Spin Resonance”</i>                                  |
| 16:05 – 16:20    | <b>Nurul Istiqomah (OP-03)</b><br><i>“Potential Fraction Cytotoxicity of Sisik Naga (Pyrrosia piloselloides [L.] M.G. Price.) Steril Fronds, Fertile Fronds, and Rhizome on Breast Cancer Cell T47D and Colon WiDr”</i> | <b>Suci Amalia (OP-07)</b><br><i>“Preparation of Monolithic Nanobiocatalyst Microreactor for Fast Protein Digestion and Their Peptide Identification by Liquid Chromatography”</i> |
| 16:20 – 16.35    | <b>Moh. Mualliful Ilmi (OP-04)</b><br><i>“Isothermal Adsorption Study of Congo Red Dye with ZSM-5 Directly Synthesized from Bangka Kaolin without Organic Template”</i>   | <b>Renny Indrawati (OP-08)</b><br><i>“Re-Evaluation on Multi-Chromatogram Approach of 3D-Chromatographic Data”</i>   |
| <b>Moderator</b> | <b>1. Dr. Hendrik O. Lintang</b><br><b>2. Dr. Yuyun Yuniati</b>   | <b>1. Rehmadata Sitepu, M.Si., Apt.</b><br><b>2. Monika Prihastyanti, M.Nat.Sc.</b>  |

Wednesday, 11 October 2017

| Time          | Ballroom   | Classroom   |
|---------------|--|---|
| 10:05 – 10:20 | <b>Septi F. Raeni (OP-09)</b><br><i>“Development of Ti<sup>4+</sup>-Immobilized Nanoporous Monolithic Polymer for Selective Separation and Detection of Phosphopeptides”</i>                               | <b>Delianis Pringgenies (OP-17)</b><br><i>“Optimal Concentration of Magrove (Rhizopora mucronata) Leaf and Propagule Based Natural Dye”</i>                                     |
| 10:20 – 10:35 | <b>Nurul Husna Sabran (OP-10)</b><br><i>“Significant Contribution of Copper(I) Pyrazolate Complex in Improving Activity of Anatase Titania”</i>  | <b>Mohamad Azani Jalani (OP-18)</b><br><i>“Size-Exclusion Liquid Chromatography for Effective Purification of Amphiphilic Trinuclear Gold(I) Pyrazolate Complex”</i>            |
| 10:35 – 10:50 | <b>Nurliana Ruslan (OP-11)</b><br><i>“Gas Chromatography with Flame Ionization and Mass Spectrometer Detectors for Evaluation of Trimethylphenol Oxidation using Heterogeneous Catalyst Nanocomposite”</i> | <b>Mohd Hayrie Mohd Hatta (OP-19)</b><br><i>“Synthesis of Highly Active Crystalline Carbon Nitride Prepared in Various Salt Melts for Photocatalytic Degradation of Phenol”</i> |
| 10:50 – 11:05 | <b>Mohamad Rafi (OP-12)</b><br><i>“Chromatographic Fingerprint Analysis for Quality Control of Medicinal Plant”</i>  | <b>Agung Bagus Pambudi (OP-20)</b><br><i>“Direct Synthesis of ZSM-5 from Kaolin without Organic Template: Part 2. Effect of Type Seeding”</i>                                   |
| 11:05 – 11:20 | <b>Yohanes Martono (OP-13)</b><br><i>“Degradation Study Of Stevioside using</i>  | <b>Florentinus D.O. Riswanto (OP-21)</b><br><i>“Analytical Method Validation and</i>  |



|                  |  |  |
|------------------|--|--|
|                  | <i>HPLC and ESI-MS/MS"</i>   | <i>Determination of Daidzein and Genistein in Ethanolic Extract of Tempeh Using RP-HPLC"</i>   |
| 11:20 – 11:35    | <b>Rahmat Budiarto (OP-14)</b><br><i>"Gas Chromatography Mass Spectrometry to Profile Leaf Metabolites of Two Popular Citrus Rootstocks in Indonesia"</i>  | <b>Edi Setiyono (OP-22)</b><br><i>"Identification of Carotenoids from Marine Bacterium Erythrobacter sp. KJ5 by Liquid Chromatography-Mass Spectrometry"</i>                         |
| 11:35 – 11:50    | <b>Leny Yuliati (OP-15)</b><br><i>"Optimization of Reaction Condition for Phenol Degradation over Platinum/Titanium Dioxide Photocatalyst"</i>   | <b>Bimo B. Santoso (OP-23)</b><br><i>"Two antimicrobial compounds drimane sesquiterpene polygodial and 11 Hydroxydrim-8-en-7-one from the stem bark of Drimys arfakensis Gibbs."</i> |
| 11:50 – 12:05    | <b>Hermin Pancasakti Kusumaningrum (OP-16)</b><br><i>"Comparison of Patchouli Oil Quality from Several Place in Batang Region, Indonesia: Attempt in Improvement of Essential Oil Quality using GC-MS Methods"</i> | <b>Mochammad Junus (OP-24)</b><br><i>"The Effect of Dairy Cattle Unit Sludge on the Nutrient of Rice Straw Composite"</i>  |
| <b>Moderator</b> | 1. Dion Notario, M.Sc.Apt.<br>2. Ruth Febriana Kesuma, S.Si., M.Si.  | 1. Rokiy Alfanaar, S.Si., M.Sc.<br>2. Eva Monica, S.Farm., M.Sc., Apt.   |

Wednesday, 11 October 2017

| Time             | Ballroom   | Classroom   |
|------------------|--|---|
| 13:00 – 13:15    | <b>Atiqa Rahmawati (OP-25)</b><br><i>"Fermentation and Purification Study of Food Grade Bioethanol from Sugar Palm Sap (Arenga Pinnata)"</i>   | <b>Sri Widarti (OP-30)</b><br><i>"The Influence of Metal Ions; Calcium, Sodium and Copper in Activating <math>\alpha</math>-Amylase with Respect to <math>\beta</math>-Cyclodextrin Grafted to Polystyrene-Diaminopropane as Stationary Phase in Affinity Chromatography"</i> |
| 13:15 – 13:30    | <b>Cheer Haan Lau (OP-26)</b><br><i>"Influence of Concentrating Method on The Quality of Orthosiphon stamineus Extract"</i>  | <b>Changi Wong (OP-31)</b><br><i>"Screening of Bioactive compounds from Nepenthes ampullaria and Nepenthes rafflesiana"</i>   |
| 13:30 – 13:45    | <b>Julius Pontoh (OP-27)</b><br><i>"Gas Chromatographic Analysis of Fatty Acid Composition in the Fresh Water Fishes in North Sulawesi"</i>  | <b>Rinaldi Idroes (OP-32)</b><br><i>"The Effect of Column and Temperature Variation on the Determination of Dead Time in Chromatographic System using Mathematical Method"</i>  |
| 13:45 – 14:00    | <b>Dewi Setyaningsih (OP-28)</b><br><i>"Development and Validation of Thin Layer Chromatography Method for Estimation of Curcumin in Dissolution Samples Containing Mixture of Curcuminoids"</i> | <b>Randi Abdur Rohman (OP-33)</b><br><i>"Role of Phenolic Acids as a Defense System in Oil Palm Infected by Ganoderma boninense"</i>  |
| 14:00 – 14:15    | <b>Hendrik O. Lintang (OP-29)</b><br><i>"Porous Kaolin-Phosphotungstic Acid Composites as Heterogeneous Catalyst for Friedel-Crafts Acylation of Anisole"</i>                                    | <b>Rosita Dwi Chandra (OP-34)</b><br><i>"Study on Pro-vitamin A and Metabolite Compounds Changes during Ripening Stages in Banana Agung Semeru (Musa paradisiaca formatypica)"</i>  |
| <b>Moderator</b> | 1. Heriyanto, S.Si., M.Si., M.Sc.<br>2. Martanty Aditya., M.Farm-Klin., Apt.   | 1. Rollando, S.Farm., M.Sc., Apt.<br>2. Renny Indrawati, S.TP., M.Nat.Sc.   |





## Porous Kaolin-Phosphotungstic Acid Composites as Heterogeneous Catalyst for Friedel-Crafts Acylation of Anisole

Hendrik O. Lintang<sup>1-3\*</sup>, Norsahika Mohd Basir<sup>4</sup> and Salasiah Endud<sup>4</sup>

<sup>1</sup>Ma Chung Research Center for Photosynthetic Pigments (MRCPP), Universitas Ma Chung, Villa Puncak Tidar N-01, Malang 65151, East Java, Indonesia

<sup>2</sup>Department of Chemistry, Faculty of Science and Technology, Universitas Ma Chung, Villa Puncak Tidar N-01, Malang 65151, East Java, Indonesia

<sup>3</sup>Centre for Sustainable Nanomaterials, IbnuSina Institute for Scientific and Industrial Research, Universiti Teknologi Malaysia, 81310 UTM Johor Bahru, Johor, Malaysia

<sup>4</sup>Department of Chemistry, Faculty of Science, Universiti Teknologi Malaysia, 81310 UTM Johor Bahru, Johor, Malaysia

\*Corresponding author: hendrik.lintang@machung.ac.id

*Keywords:* acylation, anisole, gas chromatography, porous kaolin, phosphotungstic acid

Friedel-Crafts acylations are of great importance in industry and are typically carried out by using significant amount of homogeneous acid catalysts [1]. However, the process has produced a large amount of hazardous waste, undesirable products and difficulty in separation [2]. In this work, we reported phosphotungstic acid (HPW) supported onto porous kaolin (PK) as heterogeneous catalysis PK/HPW for the acylation of anisole with propionic anhydride as an acylating agent. A series of PK/xHPW catalysts were successfully prepared with concentration of 10–40 wt% (x is the concentration of HPW) using wet impregnation method. By using pyridine adsorption-FTIR spectroscopy, acidity studies showed that porous kaolin possessed strong Lewis acid sites. In contrast, the surface acidity of the PK/HPW catalysts increased and almost comprised of strong Brønsted acid sites. The catalytic activity was evaluated using gas chromatography with flame ionization detector (GC-FID) consisting of HP-5 column (100% dimethylpolysiloxane, 25 x 0.20 mm I.D). Moreover, analysis of a product as *p*-methoxypropiofenone was carried out using GC with mass spectrometer (GC-MS) detector equipped with the same column. After 3 hours at 100 °C, the PK/30HPW catalyst (70 mg under solvent-free condition) in the mixture of anisole (30 mmol) and propionic anhydride (30 mmol) showed excellent catalytic activity with 86% conversion and high selectivity in 95% toward *p*-methoxypropiofenone. This result indicates the importance of Brønsted acid sites of well-dispersed HPW onto porous materials with balance of total pore volume and hierarchical factor value (ratio of mesoporosity to microporosity).

### References:

- [1] Spagnol, M., Gilbert, L. & Alby, D. 1996. In: J. R. Desmurs, S. Rattoy (Eds.). *The Roots of Organic Development*. Amsterdam: Elsevier.
- [2] Olah, G. A. 1973. *Friedel-Crafts Chemistry*. New York: John Wiley and Sons.




Ma Chung Research Center for Photosynthetic Pigments
 Ma Chung International Conference on Chromatoraphy

**OP-29**


**Porous Kaolin-Phosphotungstic Acid Composites as Heterogeneous Catalyst for Friedel-Crafts Acylation of Anisole**


**Hendrik O. Lintang,<sup>1,2,3</sup>**  
**Norsahika Mohd Basir<sup>4</sup>, and Salasiah Endud<sup>4</sup>**



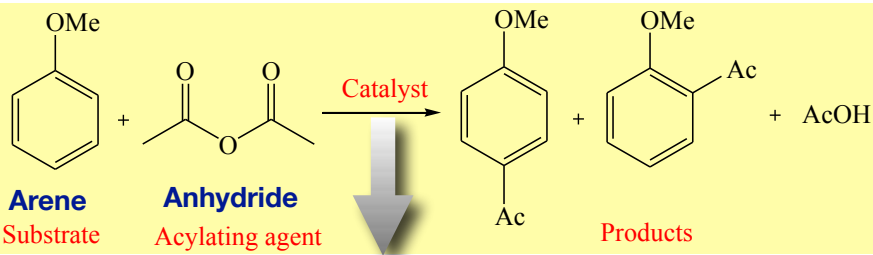
<sup>1</sup>Ma Chung Research Center for Photosynthetic Pigments (MRCPP), <sup>2</sup>Department of Chemistry, Faculty of Science and Technology, Universitas Ma Chung, INDONESIA  
<sup>3</sup>Centre for Sustainable Nanomaterials, <sup>4</sup>Department of Chemistry, Faculty of Science, Universiti Teknologi Malaysia, MALAYSIA

1



## Friedel-Crafts Acylation

**Electrophilic Aromatic Substitution Reaction between Arenes (Aromatic Rings) and Acyl Chlorides or Anhydrides using a Strong Lewis Acid Catalyst to Produce Monoacylated (Polyalkylated) Products**



**Arene**  
 Substrate

**Anhydride**  
 Acylating agent


**Products**

**Intermediates for Fine Chemical and Pharmaceutical Industries or as Precursors for Functional Polymer**


**Homogeneous Catalyst**  
**Lewis Acid Catalysts (FeCl<sub>3</sub>, ZnCl<sub>2</sub>, AlCl<sub>3</sub> and BF<sub>3</sub>)**

**Drawbacks**

- 🚫 Significant Hazardous Acids
- 🚫 High Toxicity
- 🚫 Corrosion
- 🚫 Difficult to Separate
- 🚫 Destroy the Catalysts



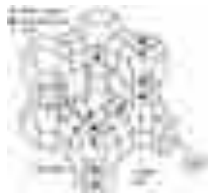
Spagnol, M., et al., 1996, Book; *The Roots of Organic Development*


2

2

# Heterogeneous Acid Catalysts

## Example of Heterogeneous Acid Catalysts



Zeolite



Clay



Mesoporous Silica



Heteropoly Acids

### Examples

- UDCat-5; 99% Selectivity  
57% Conversion
- Mordenite Zeolite; 99% Selectivity  
45% Conversion

Low Percentage of Conversion



### Advantages

- High Surface Area and Pore Volume
- Strong Acidity from Lewis Acid Sites
- Micro- and Mesoporous Structure Balance
- Easy to Separate and Recycle
- Cheaper Precursors

Sartori, G and Maggi, R., *Chemical Review* **2006**, 106, 1077

Yadav, G.D. and George, G., *Microporous and Mesoporous Materials* **2006**, 96, 36

Bhadauria, J., et al., *Journal of Chemical and Pharmaceutical Research*, **2011**, 3, 245

# Porous Clay Heterostructures (PCHs)

## Functionalization and or Modification of Clays for Increasing Porous Structures

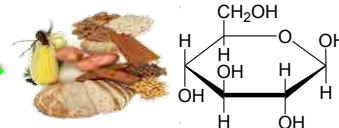


Clay

- Cationic surfactant templates
- Neutral amines co-templates



Biological Template such as Modified Starch (Glucose)



- Expensive
- Difficult to remove

Silica pillars for high thermal stability  
Porous structure with high surface area  
Combination of micro- and mesoporosity  
Modification of surface acidity (Brönsted and Lewis Acid Sites)

Galarneau, A., et al., *Nature* **1995**, 374, 239





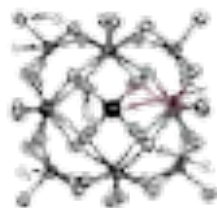
## Heteropoly Acids

A Class of Acids from the Combination of Hydrogen and Oxygen with Certain Metals and Non-Metals

Examples: Phosphotungstic acid (HPW)

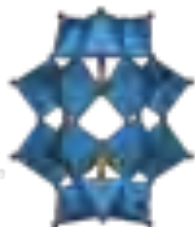


**Keggin structure**  
( $XM_{12}O_{40}^n$ )



$H_3PW_{12}O_{40}$

- Strong Brønsted acidity (pK 1.6)
- Problems in bulk form
- Low surface area
- Rapid deactivation
- Thermally less stable ( $> 200\text{ }^\circ\text{C}$ )



**Dawson structure**  
( $X_2M_{18}O_{62}^n$ )

X = P or Si  
M = Mo, W or V

**PCH-HPW Composites as Heterogeneous Catalyst**

- Good Surface Area
- Good Porosity
- Hierarchical Structure
- Acidity Properties

Campbell, K.A., et al., *Langmuir* **2002**, 21, 4738  
Parida, et al., *Langmuir* **2010**, 21, 4738

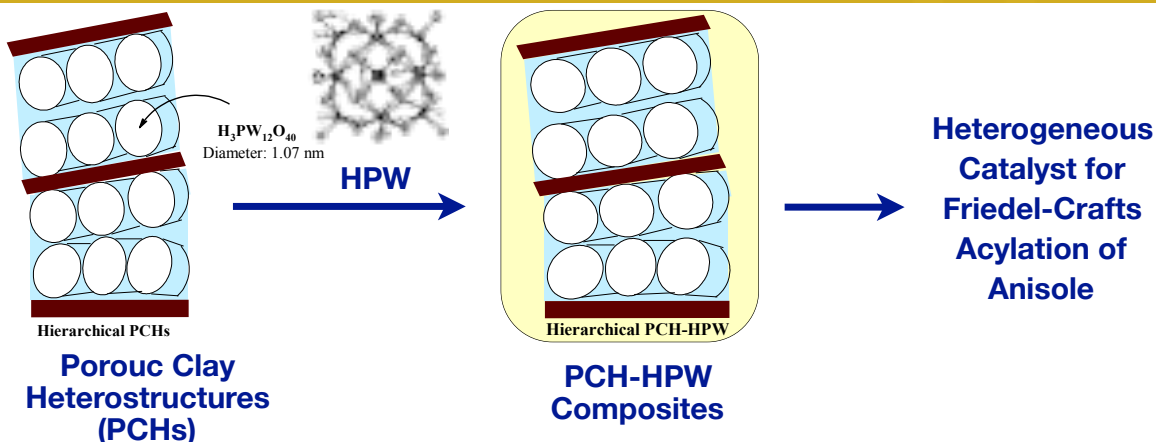


5

5



## Research Objectives



**Porous Clay Heterostructures-Phosphotungstic Acid Composites as Heterogeneous Catalyst for Friedel-Crafts Acylation of Anisole**

- Synthesis of Hierarchical Porous Clay Heterostructures (PCHs)
- Synthesis of PCHs-Phosphotungstic Acid (HPW) Composites
- Friedel-Crafts Acylation of Anisole using PCH-HPW Composites

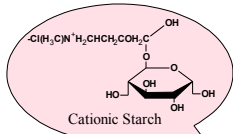


6

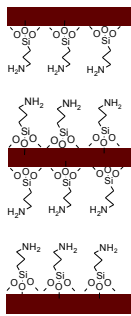
6



## Synthesis of Porous Clay Heterostructures (PCHs)

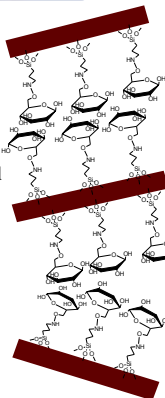


Modification of Potato Starch (PS) as Cationic Starch (CS) in 15 g/200 mL DMSO at 80 °C using NaOH (0.5 g) in water (20 mL) containing N-(3-chloro-2-hydroxypropyl)trimethylammonium chloride (12.35 mL) at 60 °C

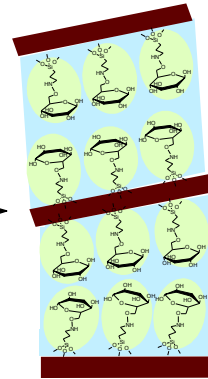


APTMS-Kaolin

Intercalation of modified starch template

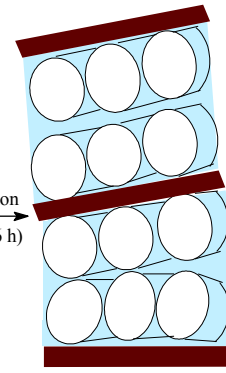


Addition of:  
(i) Dodecylamine  
(ii) TEOS  
4 h at r.t.



As-synthesized PCHs

Calcination  
500 °C (6 h)



Hierarchical PCHs  
(Porous Kaolin = PK)

modified clay/dodecylamine/TEOS = 1/ 10/ 75  
APTMS = 3-aminopropyltrimethoxysilane



Galarneau, A., et al., *Nature* **1995**, 374, 239

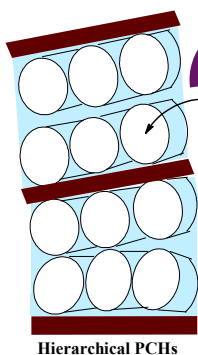
7

7



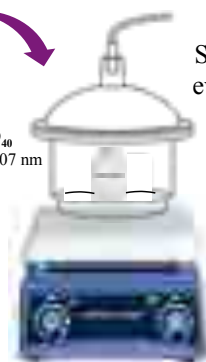
## Synthesis of PCH-HPW Composites

Wet impregnation method  
PK: 10, 20, 30 and 40 wt.% of HPW



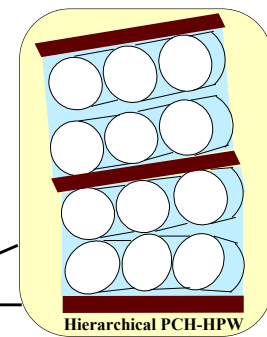
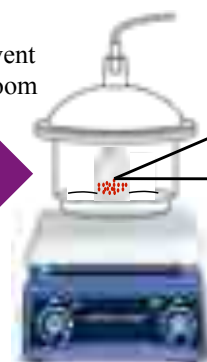
Hierarchical PCHs

$H_3PW_{12}O_{40}$   
Diameter: 1.07 nm



Stirring and solvent evaporation at room temperature

Methanol



Hierarchical PCH-HPW

PK/yHPW

y = loading amount of HPW ranging 10-40 wt%

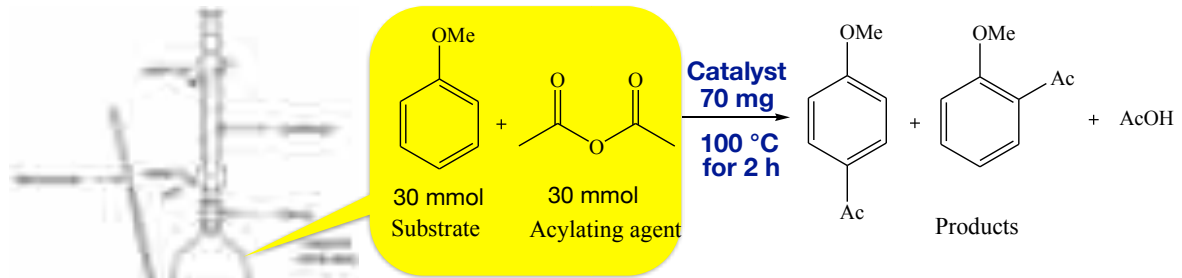


Galarneau, A., et al., *Nature* **1995**, 374, 239

8

8

## Friedel-Crafts Acylation of Anisole



### Analysis of Reaction Products:

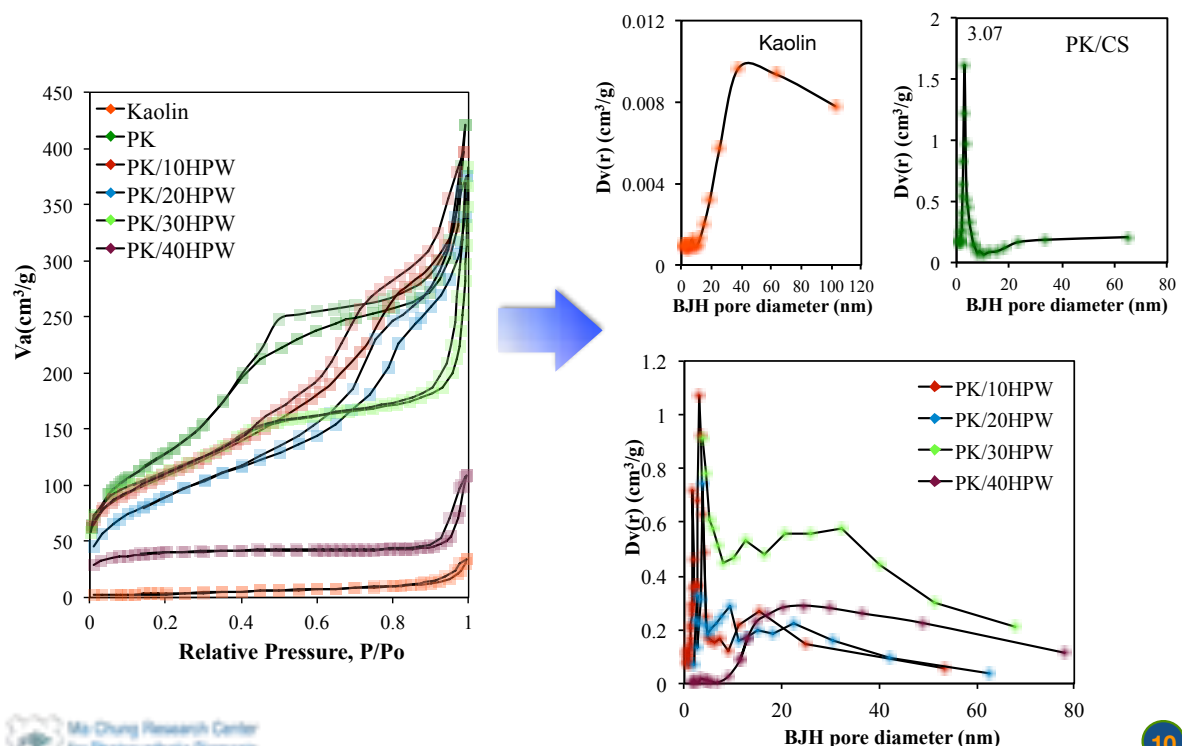
#### Gas Chromatography - Flame Ionization Detector (GC-FID)

- Hewlett Packard 5890 series II GC instrument
- HP-5 (100% dimethylpolysiloxane, 25 x 0.20 mm I.D) column
- Oven Temp. Program: initial temperature 50 °C (8 mins), rate of temperature 15 °C/min until 200 °C (5 mins).

#### Gas Chromatography - Mass Spectrometry (GC-MS)

- Agilent 6890N-5973 Mass Spectrometry
- HP-5MS column (30 m x 0.251 mm x 0.25  $\mu\text{m}$ )
- Oven Temp. Program: 50 °C (8 mins) to 200 °C (10 mins) with the rate of 15 °C/min.

## Nitrogen Absorption-Desorption of Composites





## Textural Parameters of Composites

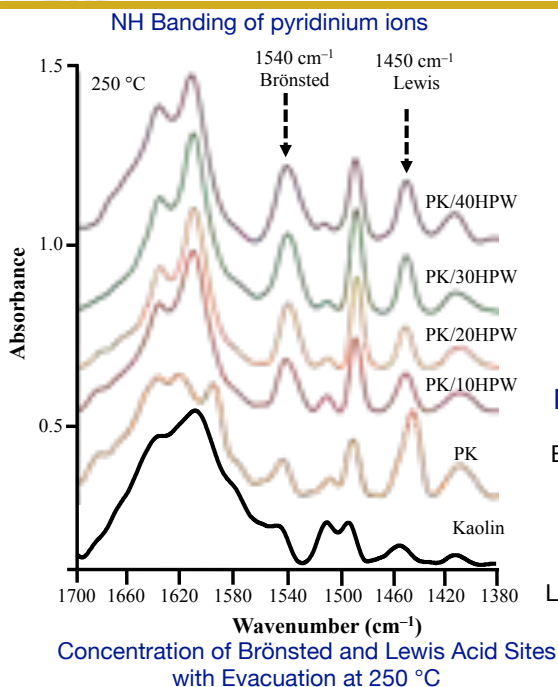
| Sample   | $S_{\text{BET}}$<br>( $\text{m}^2\text{g}^{-1}$ ) | $S_{\text{micro}}$<br>( $\text{m}^2\text{g}^{-1}$ ) | $S_{\text{meso}}$<br>( $\text{m}^2\text{g}^{-1}$ ) | $V_{\text{pores}}$<br>( $\text{cm}^3\text{g}^{-1}$ ) | $V_{\text{micro}}$<br>( $\text{cm}^3\text{g}^{-1}$ ) | $V_{\text{meso}}$<br>( $\text{cm}^3\text{g}^{-1}$ ) | $D$<br>(nm) | HF     |
|----------|---|---|--|--|--|---|-------------|--------|
| Kaolin   | 15  | -   | 23   | 0.034  | -  | 0.031   | 6.96, 38.18 | -      |
| PK/CS    | 897   | 248   | 282  | 0.653  | 0.107  | 0.546   | 3.07        | 0.0872 |
| PK/10HPW | 439   | 183   | 256  | 0.486  | 0.138  | 0.348   | 1.71, 3.03  | 0.1655 |
| PK/20HPW | 417   | 169   | 248  | 0.423  | 0.149  | 0.274   | 3.90, 9.26  | 0.2098 |
| PK/30HPW | 389   | 145   | 244  | 0.330  | 0.132  | 0.198   | 3.23        | 0.2509 |
| PK/40HPW | 184   | 68  | 116  | 0.167  | 0.052  | 0.115   | 7.80        | 0.1963 |

- Reduction of Surface Area from the Blocking of Pore Channels
- Reduction of Pore Volume from Dispersion of HPW on Surface of Mesoporous Channels
- Increasing of HF Value from High Amount of the HPW Incorporated into the Pore Structure for Reduction of Mesopore Characteristics

The hierarchy factor (HF) =  $(V_{\text{micro}}/V_{\text{pores}}) \times (S_{\text{meso}}/S_{\text{BET}})$   
 $V_{\text{micro}}$  is obtained from the t-plot method

Perez-Ramirez, J., *et al.*, *Advanced Functional Materials* **2009**, *19*, 3972

## Pyridine Absorption of Composites using FT-IR

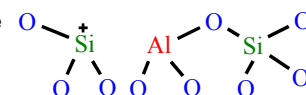


| Catalyst | Acidity ( $\mu\text{mol g}^{-1}$ ) |       | B/L | Total Acidity |
|----------|------------------------------------|-------|-----|---------------|
|          | BAS                                | LAS   |     |               |
| Kaolin   | 5.32                               | 14.48 | 0.4 | 19.80         |
| PK       | 13.32                              | 89.39 | 0.2 | 102.71        |
| PK/10HPW | 55.90                              | 34.22 | 1.6 | 90.12         |
| PK/20HPW | 80.79                              | 30.78 | 2.6 | 111.57        |
| PK/30HPW | 109.16                             | 23.08 | 4.7 | 132.24        |
| PK/40HPW | 90.25                              | 26.42 | 3.4 | 116.67        |

PK/30HPW showed the best B/L and total acidity

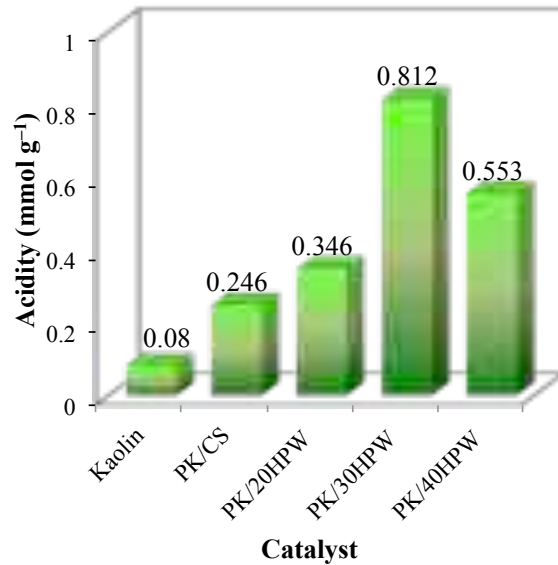
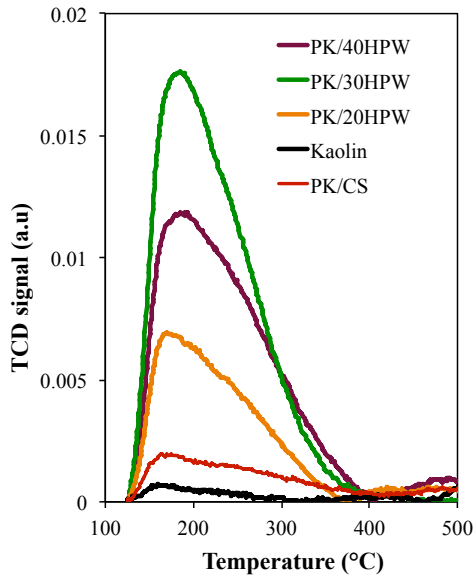
BAS = Brønsted Acid Site

LAS = Lewis Acid Site



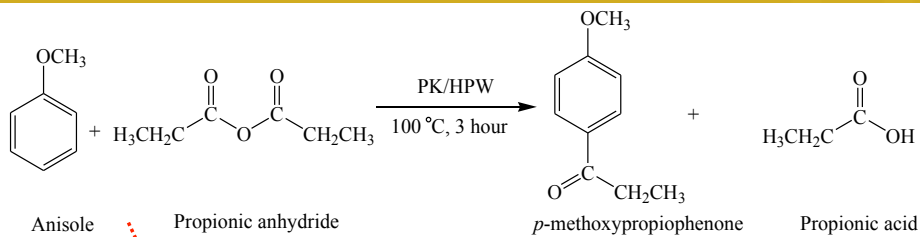
Perez-Ramirez, J., *et al.*, *Advanced Functional Materials* **2009**, *19*, 3972

## Ammonia Desorption of Composites using TPD



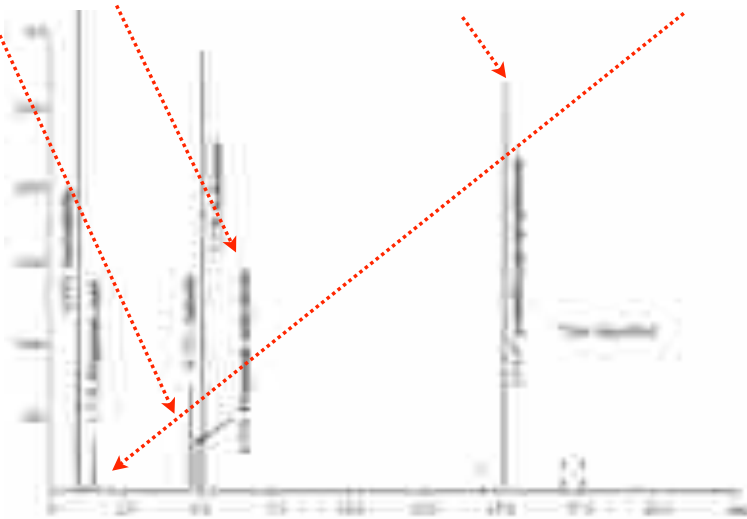
PK/30HPW showed highest acidity from the formation of monolayer coverage of HPW on the hierarchical support

## Catalytic Activity of Composites in the Acylation

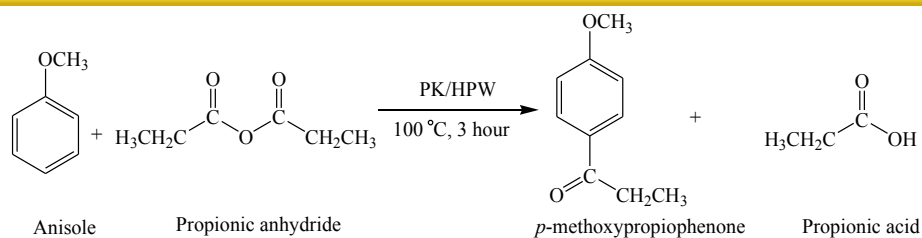


### GC-FID Chromatogram

acylation of anisole using PK/30HPW

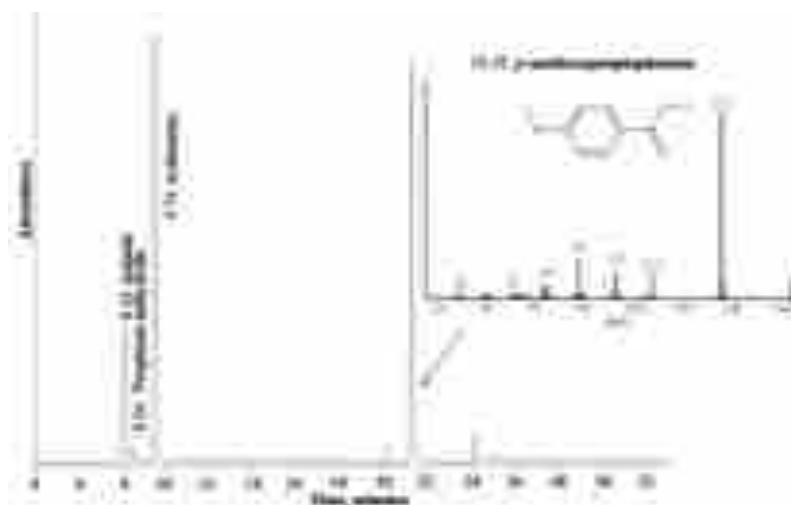


## Catalytic Activity of Composites in the Acylation

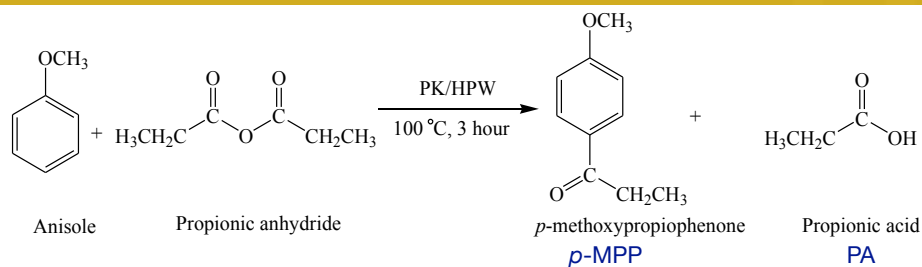


### GC-MS Spectrum

acylation of anisole using PK/30HPW



## Catalytic Activity of Composites in the Acylation

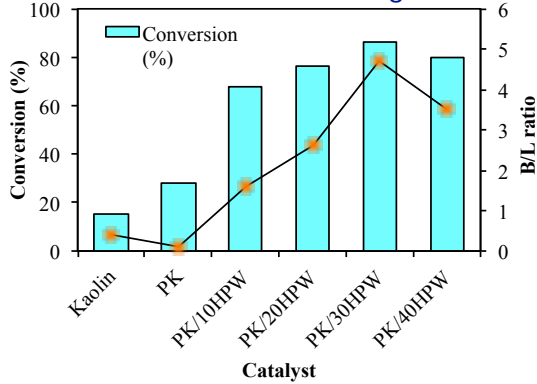


| Catalyst        | Conversion (%) | Yield of product (mmol) | Selectivity (%) |             |                |
|-----------------|----------------|-------------------------|-----------------|-------------|----------------|
|                 |                |                         | <i>p</i> -MPP   | PA          | Other products |
| HPW             | 35.98          | 10.66                   | 74.99           | 17.48       | 7.53           |
| Kaolin          | 15.15          | 4.54                    | 62.34           | 37.66       | 0              |
| PK              | 28.14          | 8.44                    | 71.95           | 27.04       | 1.01           |
| PK/10HPW        | 68.31          | 20.02                   | 81.99           | 15.44       | 2.57           |
| PK/20HPW        | 76.17          | 22.40                   | 88.98           | 8.47        | 2.55           |
| <b>PK/30HPW</b> | <b>86.30</b>   | <b>25.89</b>            | <b>95.39</b>    | <b>3.81</b> | <b>0.8</b>     |
| PK/40HPW        | 80.28          | 23.51                   | 91.16           | 7.22        | 1.62           |

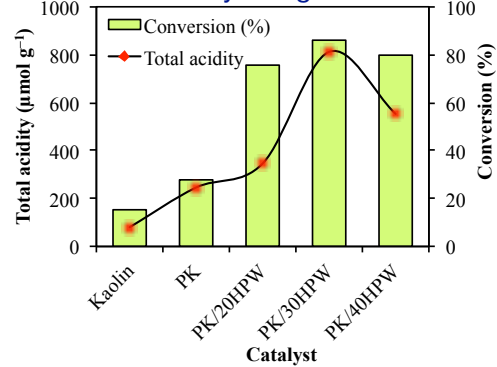


## Activity-Acidity Relationship

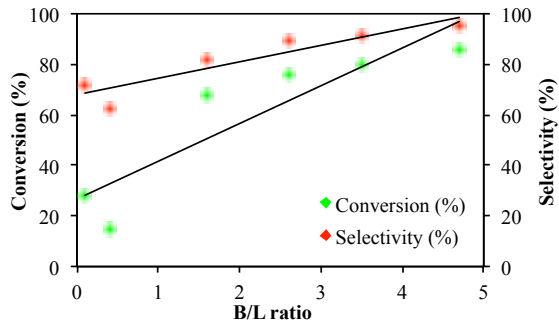
relationship of activity on the Brönsted/Lewis ratio using FT-IR



relationship of activity on the total acidity using NH<sub>3</sub>-TPD

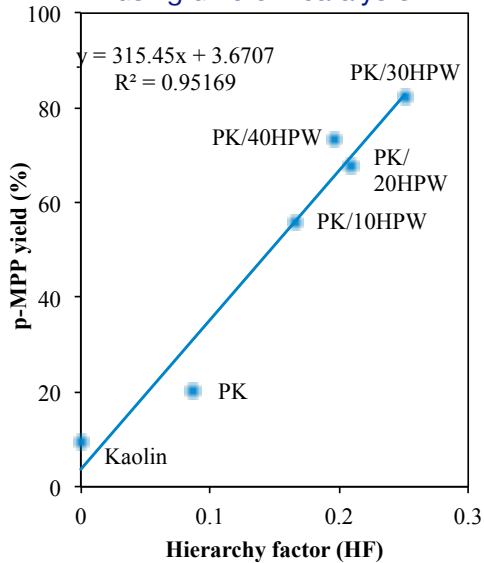


relationship of conversion and selectivity on the Brönsted/Lewis ratio using different catalysts

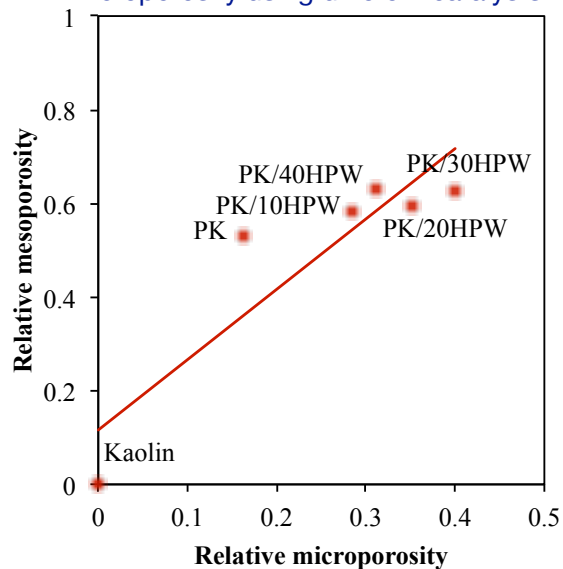


## Activity-Hierarchical Factor (HF) Relationship

relationship of activity and HF using different catalysts

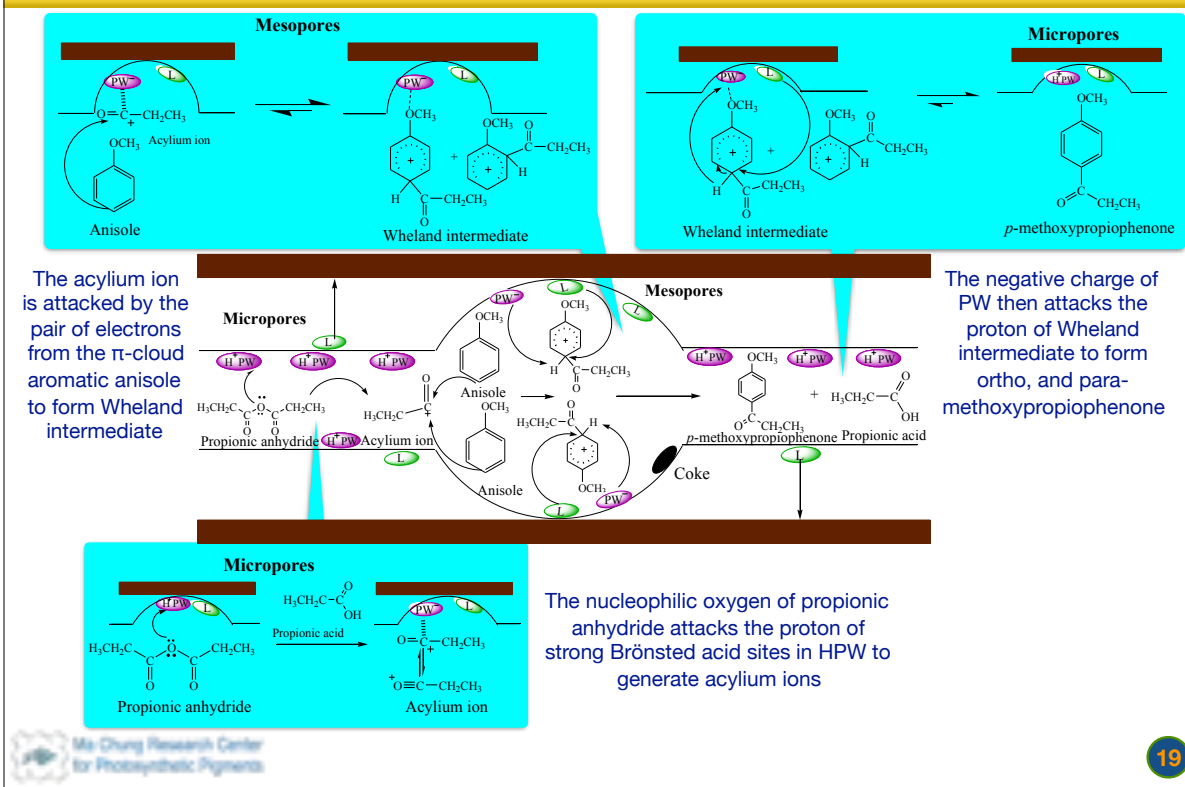


relationship of mesoporosity and microporosity using different catalysts



**The best catalyst (PK/30HPW) to give high yield of *p*-MPP has the highest HF and the relative microporosity/mesoporosity ratio**

## Proposed Mechanism of the Acylation of Anisole



## Conclusions

1. High concentration Brønsted acid sites can be generated in HPW-supported hierarchical Porous Kaolin (PK) for application as Friedel-Crafts acylation heterogeneous catalyst.
2. The 30 wt.% of HPW loading supported on hierarchical PK (PK/30HPW) was the most active catalyst in Friedel-Crafts acylation of anisole.
3. The structure-properties-catalytic activity relationship involved high concentration of strong Brønsted acid sites and hierarchical factor index of heterogeneous catalysts.



## Acknowledgements

- Centre for Sustainable Nanomaterials, Ibnu Sina Institute for Scientific and Industrial Research, Universiti Teknologi Malaysia
- Ministry of Education Malaysia for the financial support through Research University Grant Scheme (RUGS) with vote no. Q.J130000.2513.08H34 and R.J130000.2513.4F221
- Ministry of Education Malaysia for the financial support through Fundamental Research Grant Scheme (FRGS) with vote no. R.J130000.2513.4F221
- Ma Chung Research Center for Photosynthetic Pigments (MRCPP), Universitas Ma Chung, Indonesia for Financial Support as National Center of Excellence, Indonesia

